



What ice can teach us about water interactions: a critical comparison of the performance of different water models (1)



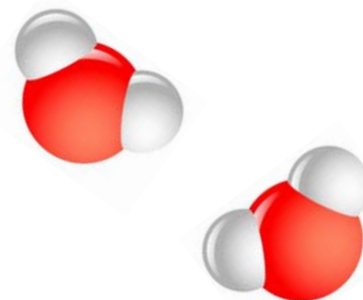
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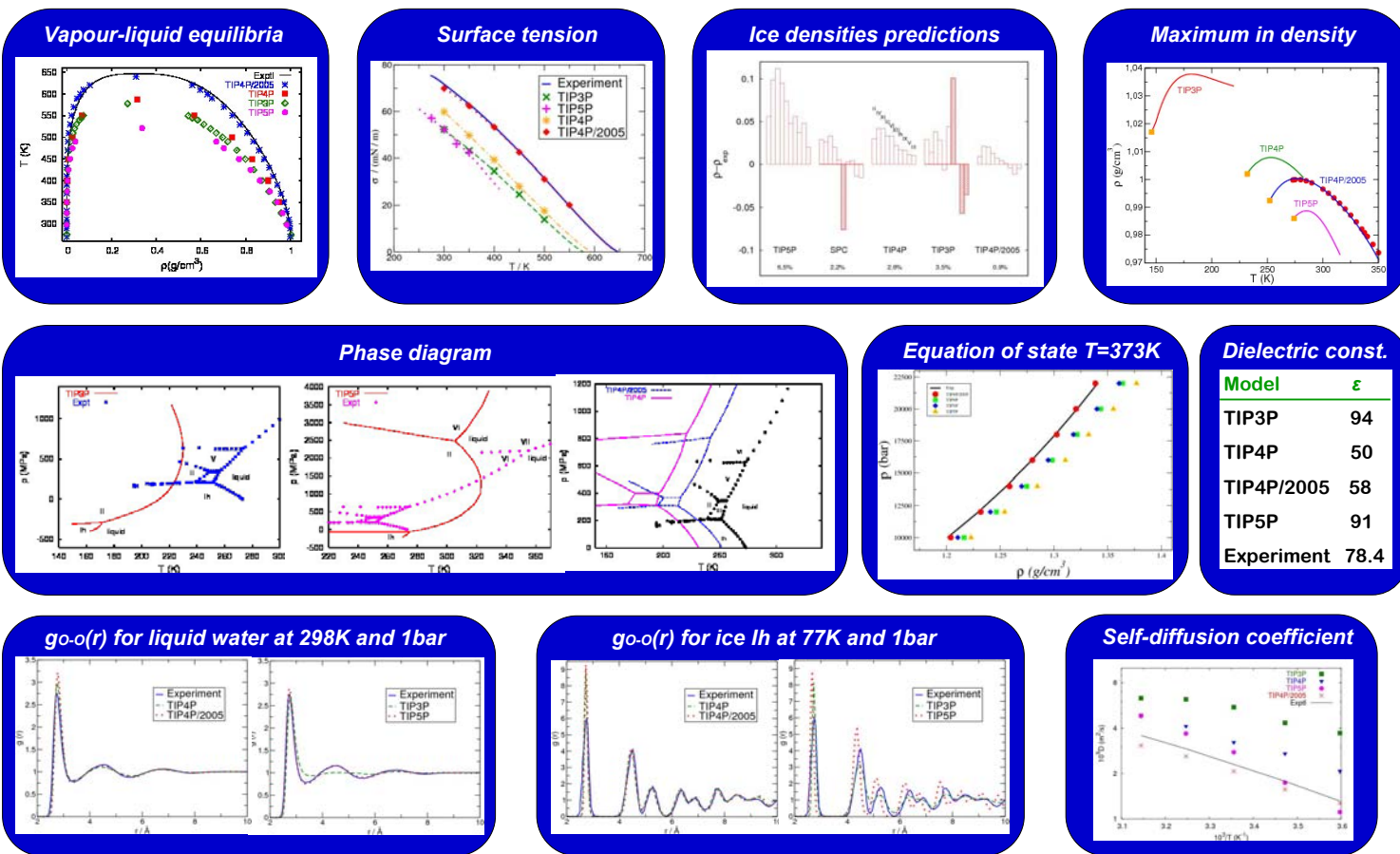
WATER POTENTIAL MODELS

POTENTIAL PARAMETERS

Model	dOH/Å	H-O-H°	$\sigma/\text{Å}$	$(\epsilon/k_B)/\text{K}$	qh/e	dom/Å	dol/Å
SPC	1.0	109.47	3.1656	78.20	0.41	0	--
SPC/E	1.0	109.47	3.1656	78.20	0.423	0	--
TIP3P	0.9572	104.52	3.1506	76.54	0.417	0	--
TIP4P	0.9572	104.52	3.1540	78.02	0.52	0.15	--
TIP4P/2005 (2)	0.9572	104.52	3.1589	93.20	0.5564	0.1546	--
TIP5P	0.9572	104.52	3.1200	80.51	0.241	--	0.70



TESTING THE MODELS FOR TEN PROPERTIES



SCORES (3=the best and 0=the worst)

Property	TIP3P	TIP4P	TIP4P/2005	TIP5P
1. VLE, T_c	1	2	3	0
2. Surface tension	1	2	3	0
3. ρ Ices	0	2	3	1
4. Phase diagram	0	2	3	1
5. T_m melting prop.	0	1	2.5	2.5
6. T_{MD} , α , K_T	0	1	3	2
7. Structure	0	1	2.5	2.5
8. EOS (high p)	2	1	3	0
9. D	0	1	3	2
10. ϵ	2	0	1	3
Total	6	13	27	14

**“For most of the properties
TIP4P/2005 yields the best
performance”**

REFERENCES

- (1) C. Vega, J.L.F. Abascal, M.M. Conde and J.L. Aragoes. *Faraday Discussions* **141** 251-276 (2009)
 (2) J.L.F. Abascal and C. Vega. *Journal of Chemical Physics* **123** 234505-1-234505-12 (2005)